

Recent aspects of glycoconjugate synthesis: A synthetic approach to the linkage region of proteoglycans*

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Abstract - A versatile synthetic route to nonsulfated, as well as mono and disulfated glycohexaosyl serines that correspond to the linkage region of proteoglycans is developed by employing a glycotriaosyl donor and a glycotriaosyl acceptor. Chemoselective removal of levuloyl groups among other ester groups without causing any acyl migration was carried out and subsequent regioselective introduction of sulfate groups was achieved successfully.

INTRODUCTION

Different repeating disaccharides of glycosaminoglycans are linked to core proteins through a common tetrasaccharide sequence "GlcA β -(1 \rightarrow 3)-Gal β -(1 \rightarrow 3)-Gal β -(1 \rightarrow 4)-Xyl β -(1 \rightarrow 3)-Ser" (ref. 1), the presence of phosphate group at O-2 of β -Xyl¹ residue in both chondroitin sulfate from the Swarm Rat Sarcoma and heparan sulfate from bovine lung was demonstrated by Oegema, Jr. and co-workers (ref. 2) in 1984 and by L.-A. Fransson and co-workers (ref. 3) in 1985, respectively.

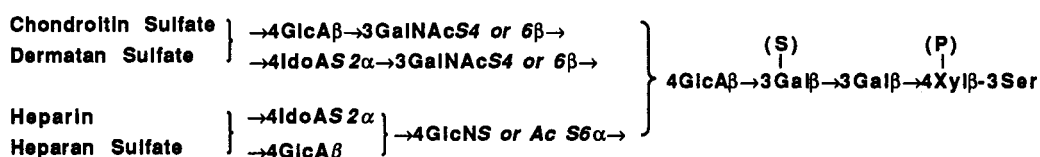


Fig. 1 Proteoglycans : Glycosaminoglycans and Linkage Regions

In 1988, Sugahara and co-workers isolated and chemically characterized neutral as well as sulfated glycohexaosylserines (1, 2 and 3) as carbohydrate-protein linkage regions of chondroitin 4-sulfate of Swarm Rat Chondrosarcoma after exhaustive enzymic digestions (ref. 4). Discovery of the presence of sulfate group at the linkage region particularly at O-4 of Gal³ of chondroitin 4-sulfate is of significant interest. The biological role played by these anionic phosphate and sulfate present in the linkage region of proteoglycans is not clear at the moment but could be the recognition signals for the transportation of the biosynthetic precursor molecules to a specific subcellular multienzymic compartment in the Golgi apparatus where specified repeating disaccharides should be assembled.

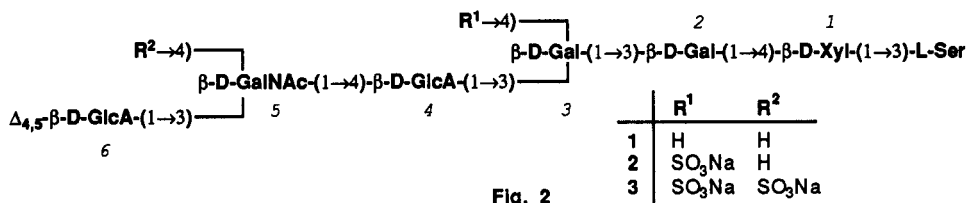
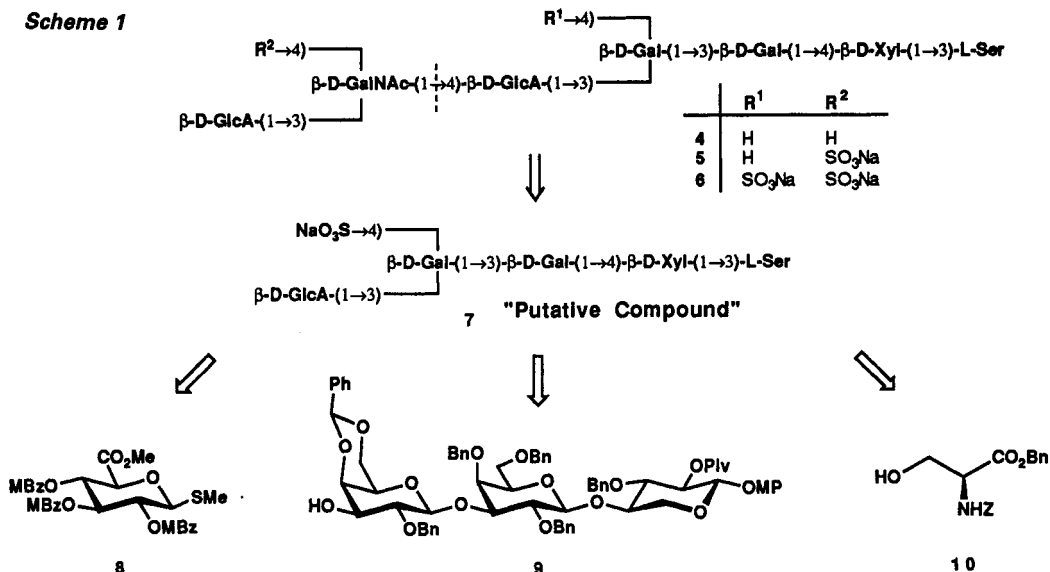


Fig. 2

* Part 90 in the series "Synthetic Studies on Cell-Surface Glycans". For part 89, see F. Goto, and T. Ogawa, *Tetrahedron Lett.*, submitted.

As part of our on-going project on the synthesis of glycosaminoglycan fragments (ref. 5), we describe here a versatile approach to the synthesis of glycohexaosyl serine 4, 5, and 6, which may be expected to function as molecular probes for the clarification of biosynthetic pathways of proteoglycans. In relevant synthetic studies elegant routes to glycotriaosyl serin have been successfully developed (ref. 6). We first describe here a synthetic route to a glycotetraosyl serine 7, a part structure of 6.

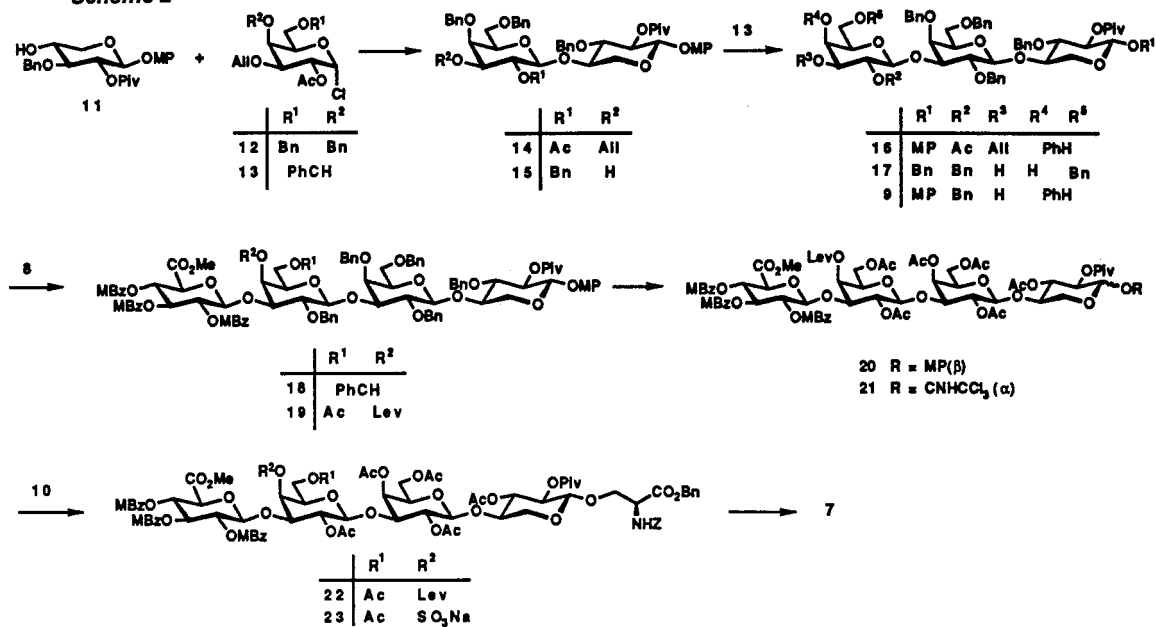
STEREOSELECTIVE SYNTHESIS OF MONOSULFATED GLYCOTETRAOSYL SERINE



Based upon a retrosynthetic analysis, we disconnected 7 into 3 parts and designed a GlcA donor 8, glycotriaosyl acceptor 9 and serine derivative 10. Thioglycoside 8 is readily available (ref. 7). Synthesis of 9 is carried out as follows. Xylopyranosyl derivative 11 (ref. 8) was coupled with 1.1 equivalents of Gal donor 12 (ref. 8) in the presence of AgOTf (ref. 9)-molecular sieves 4A (MS4A) in 5:2 toluene-CH₂Cl₂ at 20° to afford 70% of the desired β-D linked 14 (ref. 10) and 15% of the undesired α-anomer (ref. 10). 14 was converted into 15 (ref. 10) in 3 steps (1 LiOH-H₂O₂ (ref. 11), 2 BnBr, KI, Ag₂O, 3 [Ir(COD)(Ph₂MeP)₂]PF₆ (ref. 12) (Ir⁺), H₂ in THF, then I₂-H₂O, in 89% overall). Glycosylation of 15 with 1.1 equivalents of Gal donor 13 (ref. 13) in the presence of AgOTf-MS4A in CH₂Cl₂ at -23° gave 94% of β-linked trisaccharide 16 (ref. 10) which was then converted into the designed intermediate 9 (ref. 10) in 3 steps (1 LiOH, H₂O₂, 2 BnBr, Bu₄NI, NaH, 3 Ir⁺, H₂ in THF, then I₂-H₂O-NaHCO₃, 87% overall). CuBr₂-Bu₄NBr-AgOTf-MS4A (ref. 14) promoted glycosylation of 9 with 8 (1.3 equivalents) in CH₂Cl₂ at -20°~20° afforded 80% of the β-linked product 18 (ref. 10). It is to be noted that in this highly efficient glycosylation the presence of 4,6-O-benzylidene group at galactose residue (Gal)-3 is crucial to yield (1→3)-β-linked tetrasaccharide. Under the same condition, a glycosyl acceptor 17 (ref. 15) with 3,4-diol system at Gal-3 gave only 15% of the desired β-(1→3)-linked tetrasaccharide in spite of the seemingly favorable steric environment around OH-3. 18 was then converted in 3 steps (1 CSA in MeOH-CH₂Cl₂, 2 AcCl-Py, 3 Lev₂O-DMAP in Py, 86% overall) into 19 (ref. 10) which is armed with a selectively removable (ref. 16) levuloyl group at O-4 of Gal-3. Benzyl groups of 19 was replaced by acetyl groups in 2 steps (1 10% Pd-C, H₂ in EtOAc-MeOH, 2 Ac₂O-DMAP in Py, 49% overall) to afford 20 (ref. 10) which was further converted into the imidate 21 (ref. 10) in 2 steps (1 CAN (ref. 17) in MeCN-H₂O, 2 Cl₃CCN, DBU (ref. 18), 78% overall). BF₃•OEt₂ promoted glycosylation of 10 (10 equivalents) with 21 in CH₂Cl₂ at -23° gave 75% of the β-linked product 22 (ref. 10). Chemoselective removal of levuloyl group of 22 and introduction of sulfate group could be carried out efficiently to yield 23 (ref. 10) in 2 steps (1 NH₂NH₂•AcOH in toluene-EtOH (ref. 16), 2 Et₃NSO₃ (ref. 19) in DMF, 88% overall). Finally 23 was deprotected

to give the target molecule, glycotetraosyl serine **7** (ref. 10), in 3 steps (1 Pd-black, H₂ in EtOAc, 2 LiOH in 10:3 THF-H₂O, 3 NaOH in 5:1 MeOH-H₂O, 97% overall).

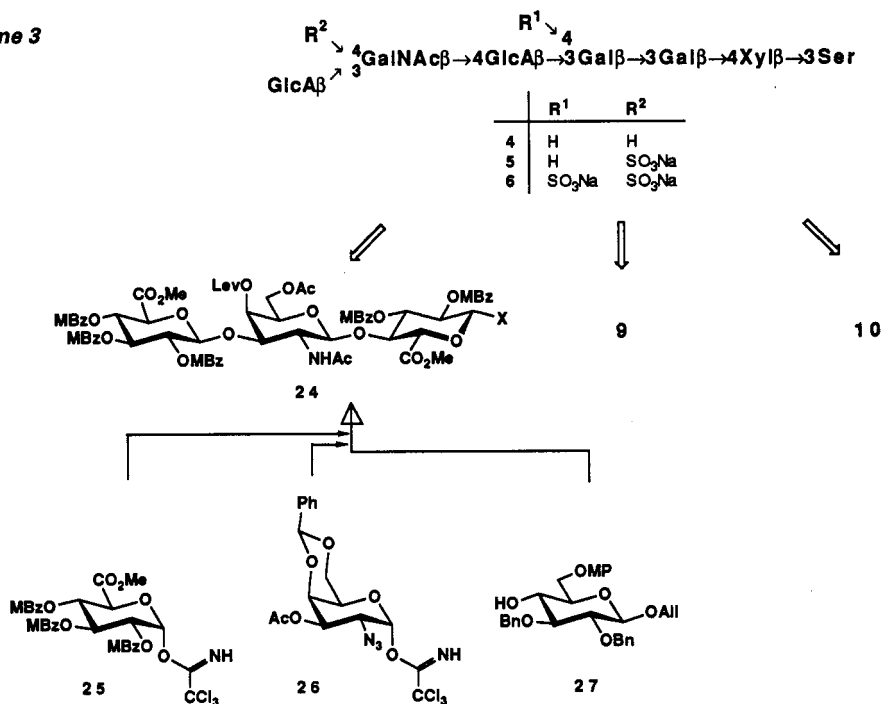
Scheme 2

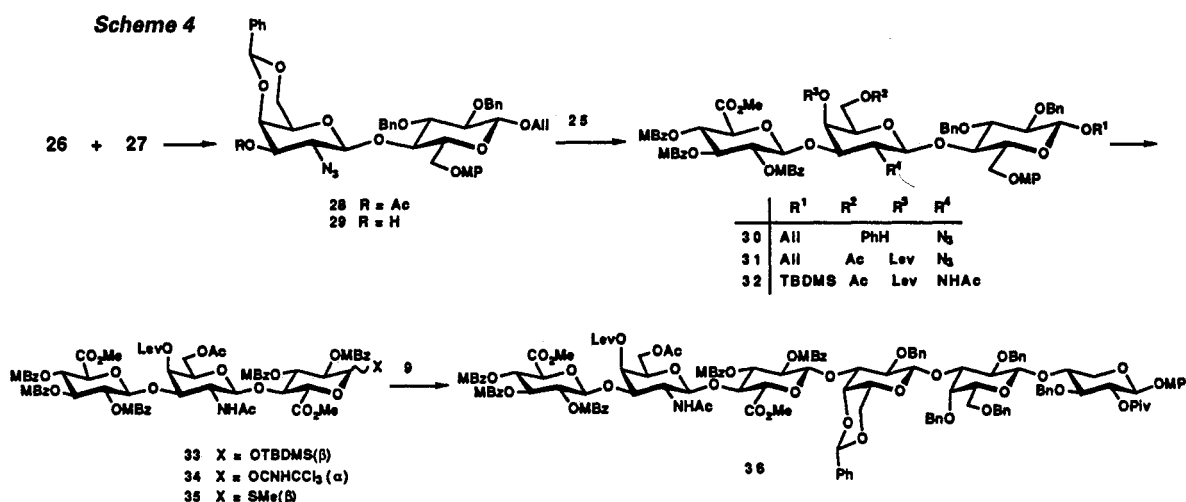


STERESELECTIVE SYNTHESIS OF MONO AND DISULFATED GLYCOHEXAOSYL SERINES

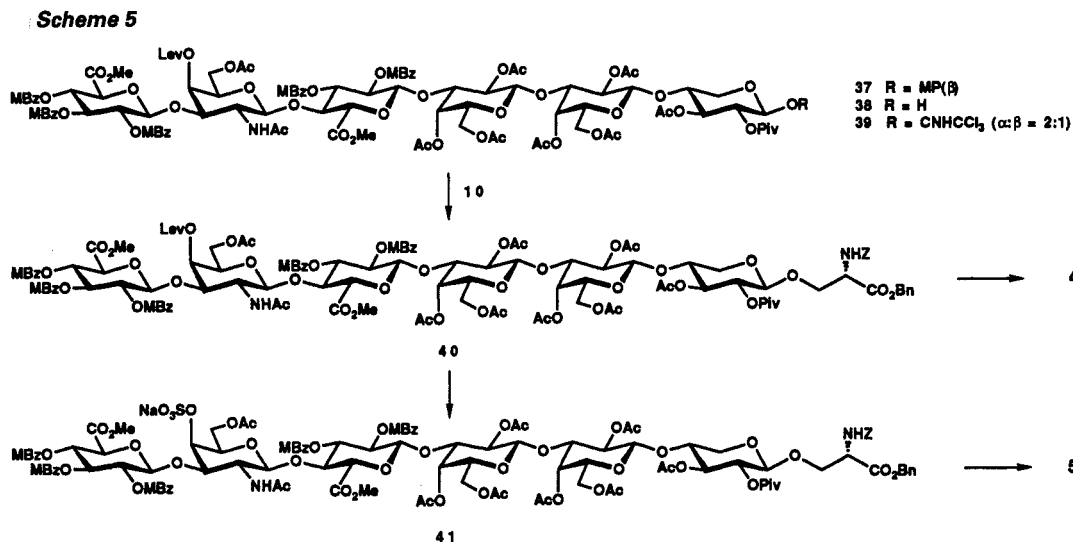
Having synthesized monosulfated glycotetraosyl serine **7** by employing a key glycotetraosyl acceptor **9**, now we have turned our attention to the use of **9** in the synthesis of glycohexaosyl serines **4**, **5**, and **6** as shown in scheme 3. Glycotetraosyl donor **24** may be obtained by successive glycosylation of **27** (ref. 20) with the imidates **26** (ref. 21) and **25** (ref. 22).

Scheme 3





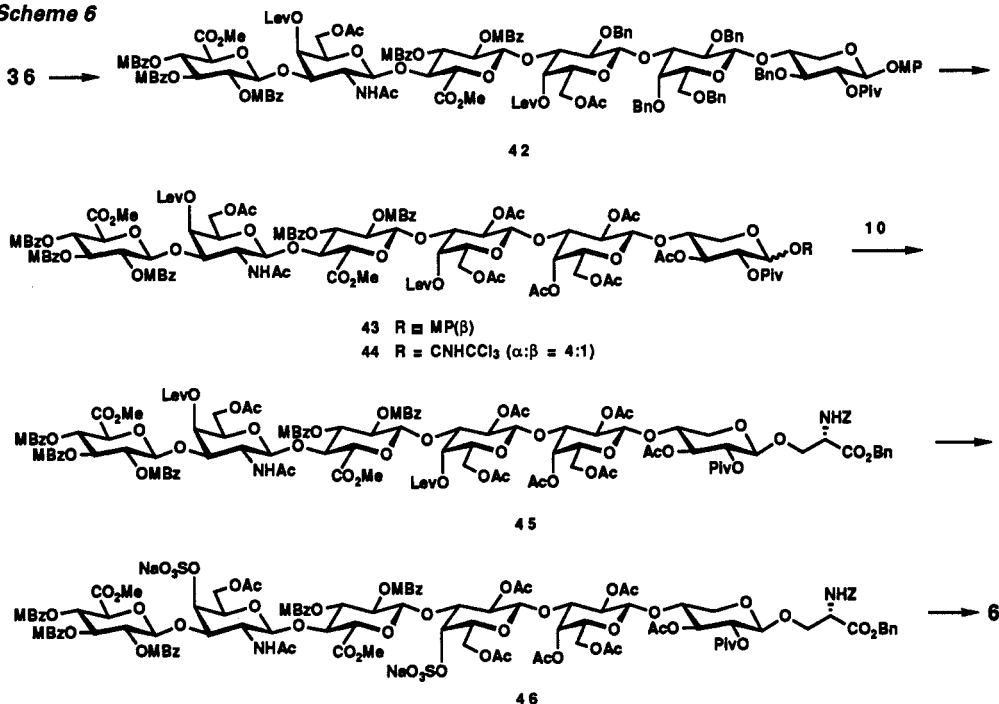
TMSOTf-MS4A promoted glycosylation of 27 with 26 (1.7 equivalents) in toluene at -78° afforded 68% of β -linked product 28 (ref. 10) and 28% of the α -anomer (ref. 10). Saponification of 28 in MeONa in 1:1 MeOH-THF to give 97% of 29 (ref. 10) which was glycosylated with 25 (1.2 equivalents) in the presence of $\text{BF}_3 \cdot \text{OEt}_2$ -MSAW300 in 30:1 toluene-(CH_2Cl_2)₂ at -25° to give stereoselectively 60% of 30 (ref. 10). No α -anomer of 30 was detected in the reaction mixture. However, the coupling between 25 and 29 (1.3 equivalents) in the presence of TMSOTf-MS4A in 200:1 toluene-(CH_2Cl_2)₂ at -23° gave 70% of a mixture of 30 and the α -anomer in a ratio of 2:5. This dramatic change of stereochemical outcome remains to be rationalized. Conversion of 30 into 31 was carried out in 3 steps (1 CSA in 1:1 MeOH- CH_2Cl_2 , 20° , 2 AcCl in Py, -5° , 3 Lev₂O, DMAP in 4:1 Py-(CH_2Cl_2)₂, 20° , 83% overall). Now we have to study further transformation of 31 into glycotriaosyl donor 24. 31 was submitted to the series of reactions, 1 $\text{HS}(\text{CH}_2)_3\text{SH}$ (ref. 22), Et₃N in MeOH, 20° , 2 Ac₂O, DMAP in Py, 3 $[\text{Ir}(\text{COD})(\text{Ph}_2\text{MeP})_2]\text{PF}_6$, H₂ in THF; then I₂-H₂O, 4 ^tBuMe₂SiCl, imidazole in DMF, to give 32 in 86% overall. Oxidative transformation of 32 was performed in 6 steps (1 10% Pd-C, H₂ in 1:1 MeOH-EtOAc, 2 MBzCl, DMAP in Py, 20° , 3 CAN in 4:1 MeCN-H₂O at 0° , 4 (COCl)₂, DMSO in CH_2Cl_2 for 5 min at -78° , then $i\text{Pr}_2\text{EtN}$ for 10 min at -20° – -15° , 5 NaClO₂ (ref. 23), NaH₂PO₄ in 2:1:1 ^tBuOH-H₂O-2methyl-2-butene 22h at 20° , 6 CH₂N₂ in EtOAc-Et₂O at 20° , 68% overall) to give 33 which was then converted into imidate 34 in 2 steps (1 Bu₄NF, AcOH in THF (ref. 24), 2 Cl₃CCN-DBU in CH_2Cl_2 , 93% overall). Thioglycoside 35 was readily prepared in 79% by treatment of 34 with Bu₃SnSMe (ref. 25) and $\text{BF}_3 \cdot \text{OEt}_2$ in CH_2Cl_2 at -25° .



Since the imidate **34** was found to give in the relevant glycosylations only low yield of the desired product upon reaction with related glycosyl acceptors, crucial coupling of glycotriaosyl acceptor **9** was examined with thioglycoside **35** (ref. 10) (0.7 equivalents) in the presence of $\text{CuBr}_2\text{-Bu}_4\text{NBr-AgOTf-MS4A}$ in CH_2Cl_2 at -25° to afford 75% of the desired glycohexaosyl product **36** (ref. 10). Catalytic hydrogenolysis of **36** in the presence of 20% $\text{Pd}(\text{OH})_2\text{-C}$ in 2:1 EtOAc-MeOH and subsequent acetylation with $\text{Ac}_2\text{O-DMAP}$ in pyridine gave 81% of **37** (ref. 10) which was further transformed into glycohexaosyl imidate **39** (ref. 10) via **38** (ref. 10) in 2 steps (1 CAN in 4:1 $\text{MeCN-H}_2\text{O}$ at 0° , 2 Cl_3CCN , DBU in CH_2Cl_2 at 0° , 90% overall). Coupling of the imidate **39** with serine derivative **10** (10 equivalents) in the presence of $\text{BF}_3\cdot\text{OEt}_2\text{-MSAW300}$ at -23° afforded 64% of **40**. The structure of **40** was confirmed by complete deprotection that gave **4** in 3 steps (1 Pd-black H_2 in 1:1 EtOAc-MeOH , 2 LiOH in 5:1 $\text{THF-H}_2\text{O}$ at -10° , 3 NaOH in 4:1 $\text{MeOH-H}_2\text{O}$, 69% overall). $^1\text{H-NMR}$ data of **4** (Fig.3) are in good agreement with the structure of **4**. Introduction of sulfate into **40** was executed in 2 steps to give **41** (1 $\text{NH}_2\text{NH}_2\cdot\text{AcOH}$ in 1:5 toluene-EtOH at 20° , 2 $\text{Et}_3\text{N}\cdot\text{SO}_3$ in DMF at 50° , 43% overall). **41** was completely deprotected in 3 steps to give monosulfated glycohexaosyl serine **5** (1 Pd-black, H_2 in 1:1 MeOH-EtOAc , 2 LiOH in 5:1 $\text{THF-H}_2\text{O}$ at -8° , 3 NaOH in 4:1 $\text{MeOH-H}_2\text{O}$ 88% overall) which gave a reasonable $^1\text{H NMR}$ as shown in Fig. 3.

To establish a route to disulfated target **6**, the key intermediate **36** was first converted in to **42** in 3 steps (1 CSA in 1:1 $\text{CH}_2\text{Cl}_2\text{-MeOH}$ at 20° , 2 AcCl in 9:4 $\text{Py-CH}_2\text{Cl}_2$ at -78° , 3 Lev_2O , DMAP in 3:1 $\text{Py-(CH}_2\text{Cl}_2)_2$ at 20° , 66% overall). **42** was converted into **44** via **43** in 4 steps (1 20% $\text{Pd}(\text{OH})_2\text{-C}$, H_2 in 2:1 EtOAc-MeOH , 2 $\text{Ac}_2\text{O-DMAP}$ in Py , 3 CAN in 2:1 $\text{MeCN-H}_2\text{O}$ at 0° , 4 Cl_3CCN ; DBU in CH_2Cl_2 at 0° , 72% overall). Coupling of **44** with **10** (10 equivalents) in the presence of $\text{BF}_3\cdot\text{OEt}_2\text{-MSAW300}$ in CH_2Cl_2 at -23° gave 64% of **45**. Simultaneous introduction of two sulfate groups in **45** was performed in 2 steps to give **46** (1 $\text{NH}_2\text{NH}_2\cdot\text{AcOH}$ in 1:5 toluene-EtOH , 2 Et_3NSO_3 in DMF at 50° , 43% overall). Finally deprotection of **46** into **6** was achieved in 3 steps (1 Pd-black, H_2 in 1:1 EtOAc-MeOH , 2 LiOH in $\text{THF-H}_2\text{O}$ at -8° , 3 NaOH in 4:1 $\text{MeOH-H}_2\text{O}$, 88% overall). $^1\text{H-NMR}$ of **6** (Fig.3) confirmed the structure.

Scheme 6



CONCLUSION

An efficient, stereocontrolled, and convergent synthetic route to the target glycohexaosyl serines **4**, **5**, and **6** was developed for the first time by employing key glycotriaosyl acceptor **9** and glycotriaosyl donor **35**. Sulfate groups could be introduced regioselectively to the specific hydroxyl groups that were temporarily protected with chemoselectively removable

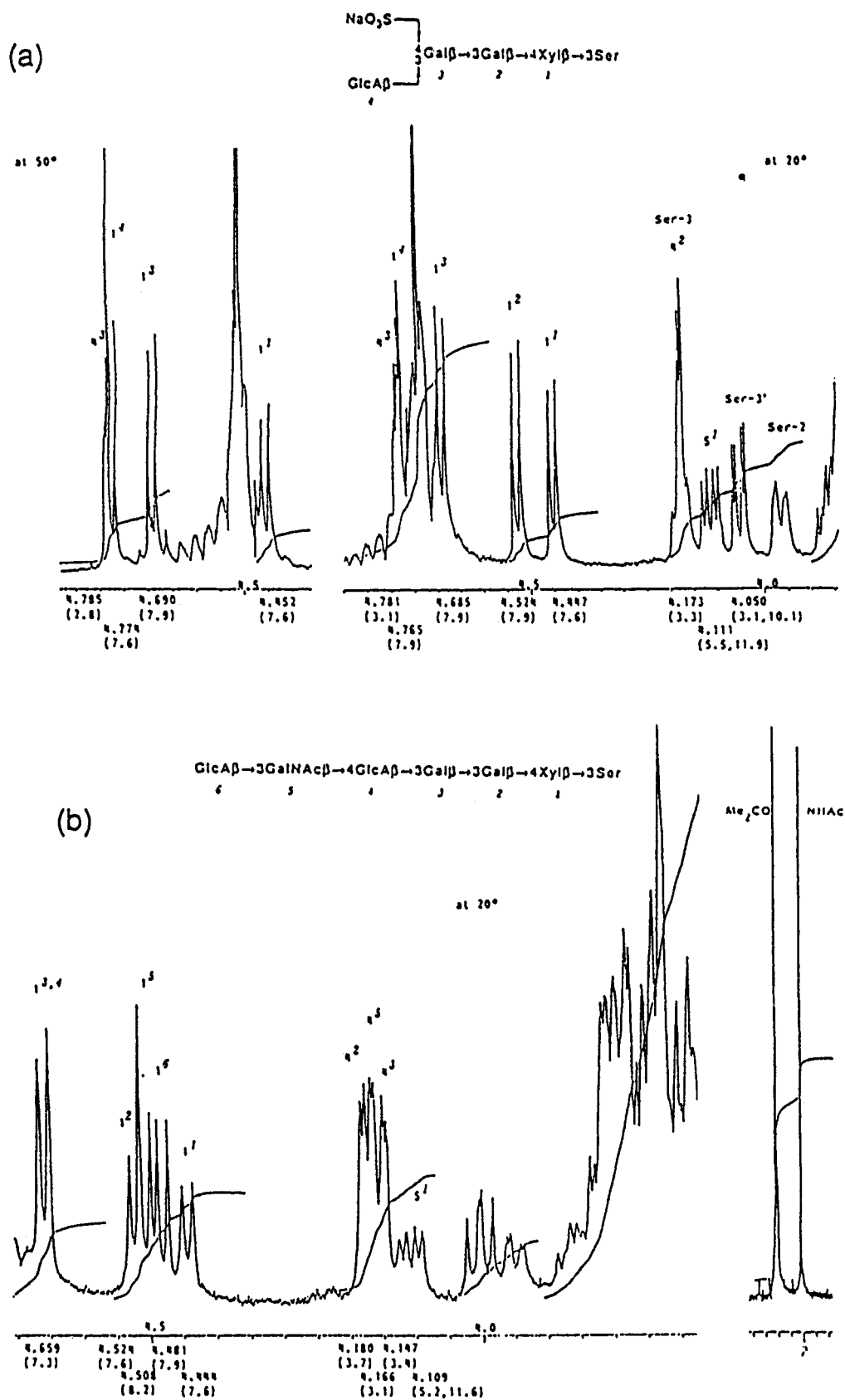


Fig. 3 500MHz ^1H NMR spectra in D_2O for compounds 7(a), and 4(b).

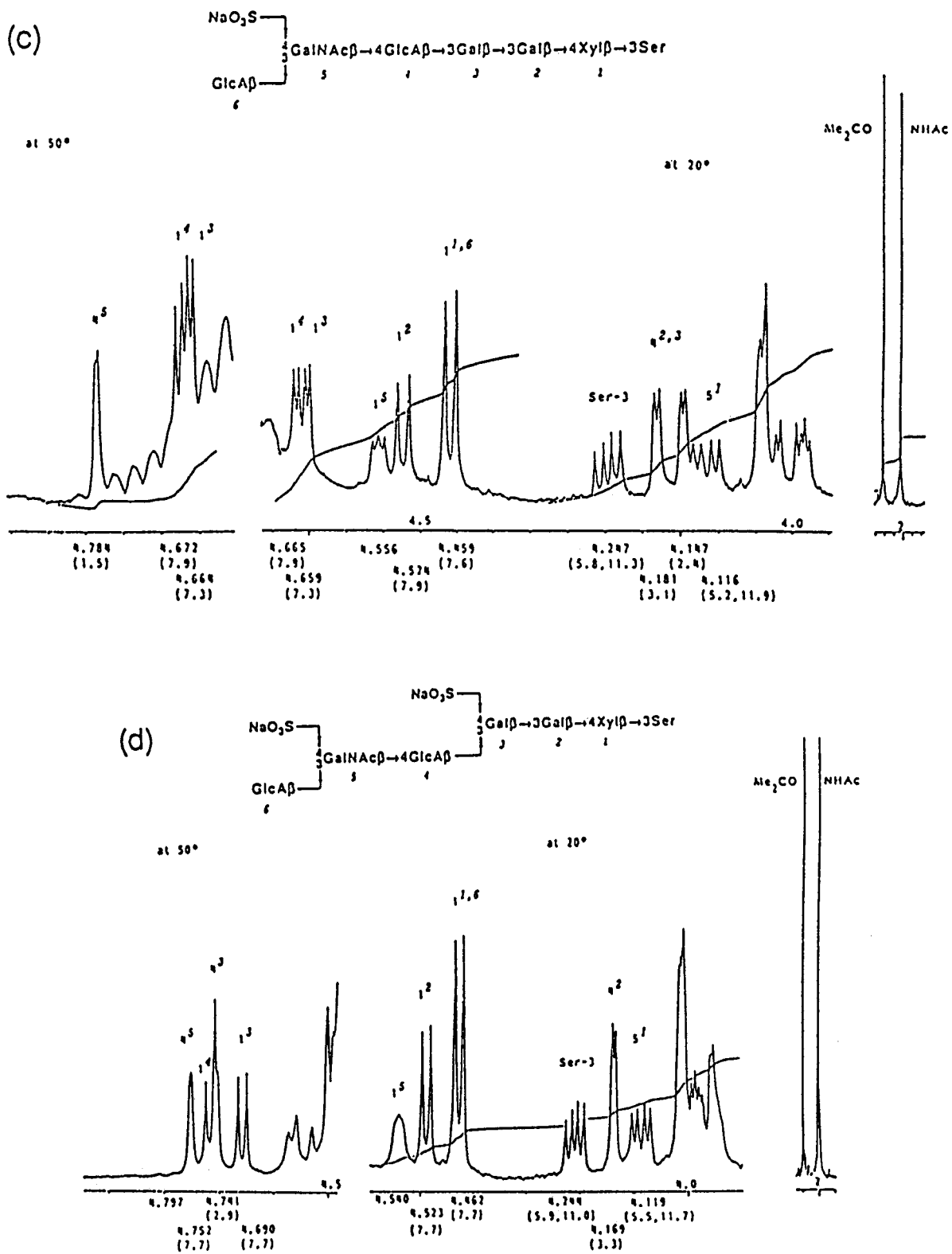


Fig. 3 (continued). 500MHz ¹H NMR spectra in D₂O for compounds 5(c), and 6(d).

levuloyl group in the presence of other ester groups. It should be noted that under the mildly basic condition employed for the removal of the levuloyl group and for the introduction of sulfate group no acetyl migration from O-6 to O-4 of Gal residue was observed.

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10. Physical data for new compounds are given below, values of $[\alpha]_D$ and δ_H , C were measured at $25^\circ \pm 3^\circ$ for solutions in CHCl_3 and CDCl_3 , respectively, unless noted otherwise. Signal assignment such as 1^3 stands for a proton at C-1 of sugar residue 3. **14**: $[\alpha]_D$ -16.4° (c 3.8); R_F 0.37 in 7:3 hexane-EtOAc; δ_H 4.868 (d, 7.0Hz, 1^1), 4.452 (d, 7.9Hz, 1^2); The a-anomer of **14**: $[\alpha]_D$ +42.9° (c 1.3); R_F 0.52 in 7:3 hexane-EtOAc; δ_H 5.268 (d, 4.0Hz, 1^2), 4.800 (d, 7.6Hz, 1^2). **15**: $[\alpha]_D$ -22.8° (c 0.4); R_F 0.25 in 7:3 hexane-EtOAc; δ_H 4.973 (d, 6.1Hz, 1^2), 4.416 (d, 7.6Hz, 1^2), 1.179 (s, Piv). **16**: $[\alpha]_D$ -27.3° (c 0.2); R_F 0.52 in 1:1 toluene-EtOAc; δ_H 5.545 (s, PhCH), 4.943 (d, 6.1Hz, 1^1), 4.901 (d, 7.9Hz, 1^3), 4.407 (d, 7.3Hz, 1^2), 1.168 (s, Piv). **9**: $[\alpha]_D$ -19.1° (c 0.9); R_F 0.40 in 1:1 toluene-EtOAc; δ_H 5.475 (s, PhCH), 4.910 (d, 6.3Hz, 1^1), 4.680 (d, 8.3Hz, 1^3), 4.361 (d, 6.9Hz, 1^2), 1.114 (s, Piv). **18**: $[\alpha]_D$ -4.7° (c 1.7); R_F 0.50 in 2:1 toluene-EtOAc; δ_H 5.631 (s, PhCH), 5.381 (d, 6.9Hz, 1^4), 4.941 (d, 5.9Hz, 1^1), 4.856 (d, 7.6Hz, 1^3), 4.370 (d, 7.3Hz, 1^2), 3.759 and 3.644 (2s, 2 OMe), 1.169 (s, Piv). **19**: $[\alpha]_D$ -1.3° (c 0.5); R_F 0.17 in 3:1 toluene-EtOAc; δ_H 5.429 (d, 4.0Hz, 4^3), 4.954 (d, 5.9Hz, 1^1), 4.820 (d, 7.9Hz, 1^3), 4.391 (d, 7.3Hz, 1^2), 3.761 and 3.704 (2s, 2 OMe), 2.202 (s, Lev), 2.003 (s, Ac), 1.166 (s, Piv). **20**: $[\alpha]_D$ +2.4° (c 1.6); R_F 0.37 in 1:2 toluene-EtOAc; δ_H 5.488 (d, 3.3Hz, 4^3), 4.908 (d, 7.3Hz, 1^4), 4.895 (d, 6.9Hz, 1^1), 4.383 (d, 8.3Hz, 1^2 and 1^3), 2.204 (s, Lev). **21**: $[\alpha]_D$ +27.2° (c 0.9); R_F 0.47 in 1:2 toluene-EtOAc; δ_H 8.630 (d, C=NH), 6.418 (d, 3.7Hz, 1^1), 5.482 (d, 3.3Hz, 4^3), 2.204 (s, Lev), 1.128 (s, Piv). **22**: $[\alpha]_D$ +2.5° (c 0.7); R_F 0.40 in 1:1 toluene-EtOAc; δ_H 5.483 (d, 3.4Hz, 4^3), 4.905 (d, 7.3Hz, 1^4), 4.377 (d, 7.9Hz, 1^3), 4.334 (d, 7.0Hz, 1^1), 4.319 (d, 7.9Hz, 1^2), 2.202 (s, Lev), 1.117 (s, Piv). **23**: $[\alpha]_D$ -14.8° (c 0.4, MeOH); R_F 0.55 in 15:1 CHCl_3 -MeOH; $\delta_H(\text{CD}_3\text{OD})$ 5.307 (d, 3.4Hz, 4^3), 5.206 (d, 7.6Hz, 1^4), 4.486 (d, 7.0Hz, 1^1), 3.657 (s, OMe), 1.115 (s, Piv). **7**: R_F 0.28 in 5:2:2:2 Me_2CO -AcOH-MeOH- H_2O ; $\delta_H(\text{D}_2\text{O})$ 4.781 (d, 3.1Hz, 4^3), 4.765 (d, 7.9Hz, 1^4), 4.685 (d, 7.9Hz, 1^3), 4.524 (d, 7.9Hz, 1^2), 4.447 (d, 7.6Hz, 1^1), 4.173 (d, 3.4Hz, 4^2). **28**: $[\alpha]_D$ +33.5° (c 1.3); R_F 0.23 in 3:1 hexane-EtOAc; δ_H 5.446 (s, PhCH), 4.511 (d, 7.6Hz, 1^1), 4.477 (dd, 3.7 and 11.0Hz, 3^2), 4.377 (d, 8.2Hz, 1^2), 3.776 (s, OMe). α -anomer of **28**: $[\alpha]_D$ +99.1° (C 1.2); R_F 0.34 in 3:1 hexane-EtOAc; δ_H 5.349 (s, PhCH), 5.900 (d, 3.7Hz, 1^2), 4.534 (d, 7.6Hz,

- 1^J), 3.770 (s, OMe). 29: [α]_D +5.5° (c 0.7); R_F 0.28 in 3:1 toluene-EtOAc; δ _H 5.516 (s, PhCH), 4.518 (d, 7.9Hz, 1^J), 4.268 (d, 8.2Hz, 1²), 3.771 (s, OMe). 30: [α]_D -6.2° (c 1.1); R_F 0.52 in 2:1 toluene-EtOAc; δ _H 5.415 (d, 7.3Hz, 1³), 4.472 (d, 7.6Hz, 1^J), 4.293 (d, 8.2Hz, 1²), 3.781 and 3.618 (2s, 2 OMe). The α -anomer of 30: [α]_D +8.7° (c 1.6); R_F 0.59 in 2:1 toluene-EtOAc; δ _H 6.033 (d, 4.9Hz, 1³), 4.478 (d, 7.6Hz, 1^J), 4.176 (d, 8.2Hz, 1²), 3.709 and 3.629 (2s, 2 OMe). 31: [α]_D -8.0° (c 0.1); R_F 0.32 in 3:1 toluene-EtOAc; δ _H 5.307 (d, 3.4Hz, 4²), 5.028 (d, 7.6Hz, 1³), 4.467 (d, 7.6Hz, 1^J), 4.284 (d, 7.9Hz, 1²), 2.171 (s, Lev), 2.005 (s, Ac). 32: [α]_D +15.4° (c 0.2); R_F 0.52 in 1:1 toluene-EtOAc; δ _H 5.404 (d, 3.7Hz, 4²), 5.402 (dd, 8.2 and 9.5Hz, 2³), 5.187 (d, 8.2Hz, 1³), 4.869 (d, 7.6Hz, 1²), 4.649 (d, 7.3Hz, 1^J), 3.779 and 3.672 (2s, 2 OMe), 2.174 (s, Lev), 1.966 (s, OAc), 1.370 (s, NAc). 33: [α]_D +11.2° (c 0.3); R_F 0.41 in 1:1 toluene-EtOAc; δ _H 3.805 and 3.644 (2s, 2 OMe), 2.220 (s, Lev), 2.049 (s, OAc), 1.548 (s, NAc). 34: [α]_D +21.6° (c 0.8); R_F 0.64 in 1:2 toluene-EtOAc; δ _H 6.720 (d, 3.7Hz, 1^J), 4.938 (d, 7.9Hz, 1²), 4.790 (d, 7.6Hz, 1³), 3.799 and 3.646 (2s, 2 OMe), 2.225 (s, Lev), 2.054 (s, OAc), 1.576 (s, NAc). 35: [α]_D +22.7° (c 0.5); R_F 0.47 in 1:2 toluene-EtOAc; δ _H 4.571 (d, 9.8Hz, 1^J), 3.806 and 3.642 (2s, 2 OMe), 2.222 and 2.202 (2s, SMe and Lev), 2.047 (s, OAc), 1.560 (s, NAc). 36: [α]_D -4.1° (c 0.7); R_F 0.48 in 1:3 hexane-EtOAc; δ _H 5.550 (s, PhCH), 5.274 (d, 6.7Hz, 1⁴), 3.770, 3.765, and 3.646 (3s, 3 OMe), 2.213 (s, Lev), 2.015 (s, OAc), 1.566 (s, NAc), 1.165 (s, Piv). 37: [α]_D +18.0° (c 0.2); R_F 0.45 in 1:4 toluene-EtOAc; δ _H 3.820, 3.764, and 3.639 (3s, 3 OMe), 2.208 (s, Lev), 1.166 (s, Piv). 38: R_F 0.20 in 1:4 toluene-EtOAc; δ _H 3.821 and 3.640 (2s, 2 OMe). 39: R_F 0.32 in 1:4 toluene-EtOAc; δ _H 8.669 (s, 0.34 H, (=NH β)), 8.626 (s, 0.66H, C=NH α), 6.410 (d, 3.3Hz, 0.66H, 1^J α), 3.820 and 3.639 (2s, 2 OMe). 40: [α]_D +11.9° (c 0.7); R_F 0.18 in 1:2 toluene-EtOAc; δ _H 3.820 and 3.639 (2s, 2 OMe), 2.206 (s, Lev), 1.112 (s, Piv). 41: [α]_D +4.8° (c 0.3, MeOH); R_F 0.39 in 10:1 CHCl₃-MeOH; δ _H(CD₃OD) 5.399 (d, 3.3Hz, 4⁴), 3.819 and 3.628 (2s, 2 OMe), 1.108 (s, Piv). 4: R_F 0.40 in 4:4:5:3 Me₂CO-MeOH-AcOH-H₂O; FAB MS m/z 1159 [M-Na]⁻, 1137 (M+1-2Na)⁻, 1115 [M+2H-3Na]⁻. 5: R_F 0.35 in 4:4:5:3 Me₂CO-MeOH-AcOH-H₂O; δ _H (D₂O, 50°), 4.784 (d, 1.5Hz, 4⁵); δ _H(D₂O, 20°), 4.665 (d, 7.9Hz, 1⁴), 4.524 (d, 7.9Hz, 1²), 4.459 (d, 7.6Hz, 1^J and 1⁶). 42: [α]_D +1.7° (c 1.5); R_F 0.63 in 1:4 hexane-EtOAc; δ _H 5.331 (d, 3.0Hz, 4³), 5.247 (d, 3.3Hz, 4⁵), 2.223 and 2.191 (2s, 2Lev), 1.165 (s, Piv). 43: [α]_D +13.8° (c 0.9); R_F 0.20 in 1:4 toluene-EtOAc; δ _H 3.821, 3.766, and 3.643 (3s, 3 OMe), 2.209 and 2.203 (2s, 2Lev), 1.167 (s, Piv). 44: R_F 0.29 in 1:6 toluene-EtOAc; δ _H 8.670 (s, 0.2H, C=NH β), 8.628 (s, 0.8H, C=NH α), 6.413 (d, 3.6Hz, 0.8H, 1^J α), 3.819 and 3.640 (2s, 2 OMe). 45: [α]_D +6.8° (c 0.4); R_F 0.25 in 1:4 toluene-EtOAc; δ _H 3.820 and 3.641 (2s, 2 OMe), 2.208 and 2.198 (2s, 2Lev), 1.112 (s, Piv). 46: [α]_D +6.5° (c 0.5, MeOH); R_F 0.49 in 7:1 CHCl₃-MeOH; δ _H(CD₃OD) 3.730 and 3.671 (2s, 2 OMe), 1.111 (s, Piv). 6: R_F 0.32 in 4:4:5:3 Me₂CO-MeOH-AcOH-H₂O.
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