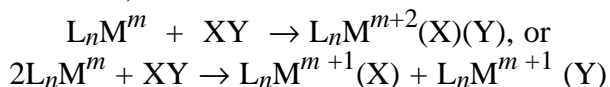
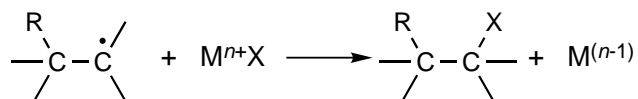
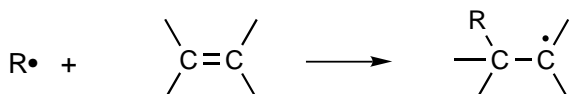
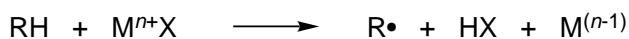


oxidative addition

The *insertion* of a metal of a metal complex into a covalent bond involving formally an overall two-electron loss on one metal or a one-electron loss on each of two metals, i.e.



In *radical* chemistry, the term is used to indicate a radical addition to a carbon-carbon double bond, under oxidative conditions. For example:



1994, 66, 1149