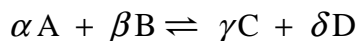


mass-law effect

At equilibrium, the product of the activities (or concentrations) of the reacting species is constant. Thus for the equilibrium:



$$K = [C]^\gamma [D]^\delta / [A]^\alpha [B]^\beta$$

See also *common-ion effect, equilibrium*.

1994, 66, 1138