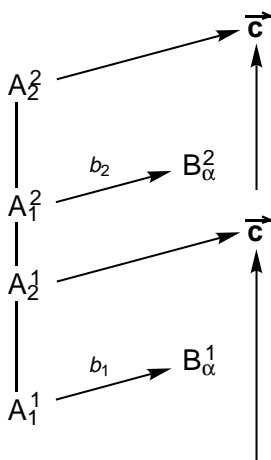


### isomorphous structures (in polymers)

In the crystalline state, polymer chains are generally parallel to one another but neighbouring chains of equivalent conformation may differ in chirality and/or orientation.

Chains of identical *chirality* and *conformation* are isomorphous. Chains of opposite chirality but equivalent conformation are enantiomorphous. For example, two ...TG<sup>+</sup>TG<sup>+</sup>TG<sup>+</sup>... helices of isotactic poly(propylene) are isomorphous. Isotactic poly(propylene) chains of the ...TG<sup>+</sup>TG<sup>+</sup>TG<sup>+</sup>... and ...G<sup>-</sup>TG<sup>-</sup>TG<sup>-</sup>T... types are mutually enantiomorphous.

With regard to orientation, consider a repeating side group originating at atom A<sub>1</sub><sup>i</sup>, the first atom of the side group being B<sub>α</sub><sup>i</sup>. For certain chain symmetries (helical, for instance) the bond vectors  $\vec{b}(A_1^i, B_\alpha^i)$  have the same components (positive or negative)  $\vec{b} \cdot \vec{c} / |\vec{c}|$  along the *c* axis for every *i*.



Two equivalent (isomorphous or enantiomorphous) chains in the crystal lattice, having identical components of the bond vectors along *c*, both positive or both negative, are designated *isoclined*; two equivalent chains having bond vectors along *c* of the same magnitude but opposite sign are designated *anticlined*.

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