

Hartree energy

Atomic fundamental physical constant used as *atomic unit* of energy: $E_h = \hbar^2/m_e a_0^2 = 4.359\,7482\,(26) \times 10^{-18}$ J, where \hbar is the Planck constant divided by 2π , m_e the electron rest mass and a_0 the Bohr radius.

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